

Note

COMMENTS ON "KINETIC STUDIES ON SOME RARE EARTH CHELATES OF LAWSONE USING THERMOGRAVIMETRIC CURVES"

S. D. Kapila, K. K. Pathak and B. M. L. Bhatia

In a quite recent short communication which appeared in this journal (1), S. D. Kapila and coworkers applied the Coats-Redfern equation (2) to obtain the values of the activation energy for the nonisothermal decompositions of some rare earth chelates. This equation was given in the form:

$$\log \left[-\log \frac{1-\alpha}{T^2} \right] = \log \frac{AR}{a_E} \left[1 - \frac{2RT}{E} \right] - \frac{E}{2.303 RT}$$

which is wrong. The right form of the equation is:

$$\log \left[\frac{-\ln(1-\alpha)}{T^2} \right] = \log \frac{AR}{aE} \left[1 - \frac{2RT}{E} \right] - \frac{E}{2.303 RT}$$

It could be merely that the equation was written in an erroneous form, the detailed calculation being correct. However, the authors should check the correctness of the activation energy values.

References

- 1 S. D. Kapila, K. K. Pathak and B. M. L. Bhatia, *J. Thermal Anal.*, 29 (1984) 1393. 2 A. W. Coats and J. P. Redfern, *Nature*, 211 (1964) 68.

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